Eco-friendly hydrogen peroxide elimination by REDUZOL RED



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What is **REDUZOL RED**?

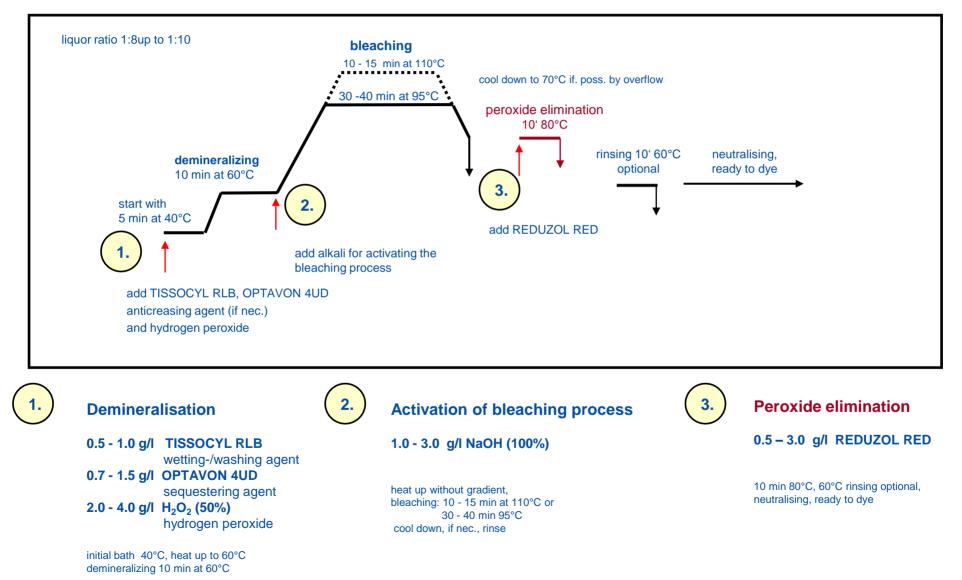
- high efficient reducing agent for peroxide elimination after hydrogen peroxide bleaching
- chemical base: combination of inorganic and organic reducing agents
- active matter about 60%

The advantages of REDUZOL RED:

- complete elimination of residual peroxide after bleaching due to synergistic effects getting from the combination of inorganic and organic reducing agents
- · increasing of reproducibility of reactive dyeing after pre-bleaching
- easy handling because of the liquid formulation
- dosable, pumpable
- approval for GOTS 3.0 in progress

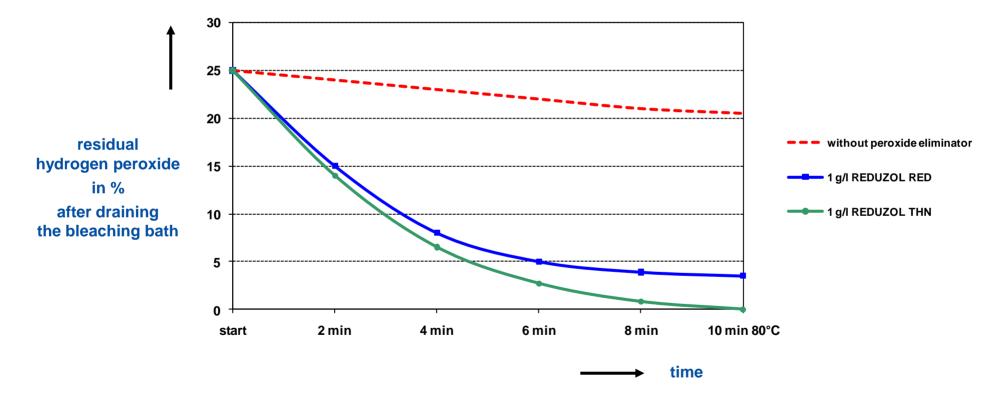


Process scheme for OPTABLEACH:





Peroxide elimination after OPTABLEACH bleaching processes, determined by titration with potassium permanganate:



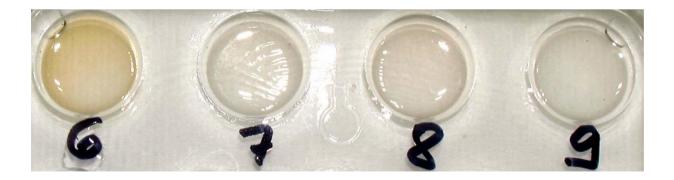
REDUZOL RED shows an excellent reducing effect !

The highly concentrated version REDUZOL THN has a slightly higher redox potential than REDUZOL RED, but it has to be dissolved before use since it is a salt.



Determination of hydrogen peroxide by spot tests with titanyl chloride:

- A quantitative determination of hydrogen peroxide in bleach liquors can be carry out easily and rapidly by using colorimetric methods.
- The method depends on the yellow-orange colour produced by the reaction of peroxide with a titanium compound. Testing: After peroxide elimination 1-3 drops of the titanium reagent are added to the liquor. After 1 minute it can be judged if a yellow colour is produced or not.
- Evaluation of the liquor after peroxide elimination:
 - deep orange colour => high amount of residual peroxide is still left,
- yellowish to colourless => peroxide is eliminated sufficiently



Description:

6) without peroxide eliminator
7) 1.0 g/l REDUZOL THN
8) 0.5 g/l REDUZOL RED
9) 1.0 g/l REDUZOL RED